

Synthesis of some thiazolidinones and N-acetyl amino derivatives  
from 4-amino sulphamethaoxazole

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**Abstract**

4-amino sulphamethaoxazole was reacted with chloro acetyl chloride to convert the amino group to 4-chloro acetyl amino sulphamethaoxazole (compound 1).The amide was then allowed to react with potassium thiocyanate to prepare 4-(2-imino-4-oxo-thiazolidinyl) compound(2) which contained thiazolidinone ring. Some aromatic amines were reacted with 4-chloro acetyl amino sulphamethaoxazole compound (1) to prepare 4-substituted anilino acetyl amino sulphamethaoxazole compounds (1A-H). 4-(2-imino-4-oxo-thiazolidinyl) Compound (2) was reacted with different aromatic aldehydes to prepare 4-(5-arylidene-2-imino-4-oxo-thiazolidinyl sulphamethaoxazole) compounds (2A-H). The prepared compounds were identified and for the prepared compounds such as (1D,1H,2,2B,2E) <sup>1</sup>H-n.m.r spectra was used.

**Keywords:** 4-amino sulphamethaoxazole, thiazolidinone, N-acetyl amino compounds.

**الملخص**

تم مفاعلة المركب 4-أmineo سلفاميثاوكسازول مع كلورو أسيتاييل كلورايد لتحويل مجموعة الامين الى مجموعة أمайд كما في المركب (1) [4-كلورو أسيتاييل أmineo سلفا ميثاوكسازول] وهذا المركب الحاوي على مجموعة أمайд تم مفاعلته مع ثيوسيانات البوتاسيوم لتحضير المركب (2) [4-(2-إmineo-4-أوكسو ثايزوليدينيل) سلفاميثاوكسازول] الحاوي

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على حلقة ثيازوليدينون. وتم مفاعلة بعض الامينات الاوروماتية مع المركب (1) لتحضير المركبات (1A-H) [4-4-أريليدين-2-إيمو-4-أوكسو ثيازوليدينول] . وتم مفاعلة المركب (2) مع الديهايدرات أروماتية مختلفة لتحضير المركبات (2A-H) [5-أريليدين-2-إيمو-4-أوكسو ثيازوليدينول] سلفا ميثا أوكسوزول []. وتم تشخيص المركبات المحضرة باستخدام مطيافية الاشعة تحت الحمراء وقياس درجة الانصهار ومطيافية الرنين النووي المغناطيسي-البروتون بعض المركبات المحضرة (1D,IH,2,2B,2E).

**الكلمات المفتاحية:** مركبات 4-أمينو سلفاميثاوكساسول، ثيازوليدينون، N-استيل امينو.

### Introduction

Thiazolidine ring is of considerable interest as it is a structure in various synthetic pharmaceuticals displaying a broad spectrum of biological activities<sup>(1-5)</sup>. Many compounds of 5-arylidine-2-imino-4-thiazolidinone have been prepared by ameya and et.al<sup>(6)</sup>. Glyoxilic acid was reacted with cysteine to prepare thiazolidine- 2,4-dicarboxylic acid that used as a ligand with divalent and trivalent metal ion<sup>(7-8)</sup>. Derivatives of thiazolidinone have been prepared by different methods and chemical reagents<sup>(9-15)</sup>. Oxazole which are part of structures of prepared compounds are known to exhibit interesting biological activities<sup>(16-20)</sup>. Oxazoles have been demonstrated to be very versatile building blocks in organic synthesis<sup>(21-28)</sup>. So many methods were used to prepare thiazolidinone derivatives<sup>(29-32)</sup>.

### Experimental

#### **Materials**

All Materials were from Aldrich and were used further purification.

#### **Instruments**

- 1- FT-IR Spectrophotometer model Shimadzu 8400 , [400-4000 cm<sup>-1</sup>].
- 2- Melting Point Apparatus model Gallenkamp.
- 3- <sup>1</sup>H-n.m.r. 300 MHz Bruker 2003 Jordan in DMSO-d6.

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**Synthesis of 4-chloro acetyl amino sulphamethaoxazole(compound 1)**

To astirred solution of 4-amino sulphamethaoxazole (0.01mole,1.26g) and triethyl amine (0.01mole,1.02ml) in dioxine (50ml),mono chloro acetyl chloride (0.01mole,1.13ml) was added dropwise. The reaction mixture was refluxed for 12 h. the excess of solvent was evaporated. The solid obtained was washed with water, filtered, dried and crystallized from ethanol<sup>(6)</sup>.

**Synthesis of 4-substituted anilino acetyl amino sulphamethaoxazole (compound1A-H).**

A mixture of [compound1 (0.1mole,32.95g)] and the substituted aromatic amine (0.1mole) in ethanol (30ml) was refluxed for 6h. after cooling the resulting solid was filtered, dried and crystallized from 80% ethanol<sup>(29)</sup>.

**Synthesis of 4-(2-imino-4-oxo-thiazolidinyl) sulphamethaoxazole (compound 2).**

A mixture of [compound1(0.01mole,3.29g)],potassium thiocyanate(0.02mole, 1.94g) and acetone (50ml) is refluxed for about 3h. excess of solvent is removed and the residue is stirred with water(50ml). The solid product is filtered washed with water, dried and crystallized from ethanol<sup>(6)</sup>.

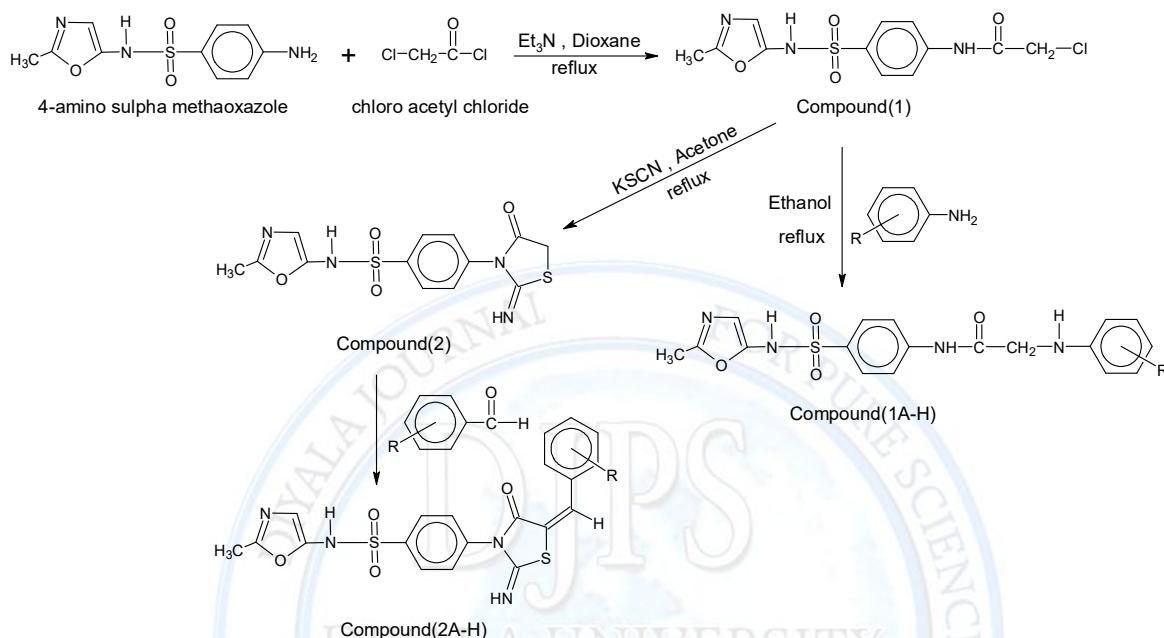
**Synthesis of 4-(5-arylidene-2-imino-4-oxo-thiazolidinyl sulphamethaoxazole)  
(compounds 2A-H).**

The substituted aromatic aldehyde(0.02mole) and [compound 2 (0.01mole, 3.52g)] are added to a solution of anhydrous sodium acetate (0.02mole) in acetic acid(30ml).The mixture is refluxed for 5h. and cooled to room temperature. The solid product is filtered, washed with water, dried and crystallized from methanol<sup>(6)</sup>.

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**Scheme 1: Path ways for prepared compounds**



**Results and Discussion**

4-Amino sulpha metha oxazole was reacted with chloro acetyl chloride to prepare compound(1) in which amino group converted to amide group infra red data showed that disappearing of amino group at 3460cm<sup>-1</sup> and appearing of –NH at 3234cm<sup>-1</sup>,-CONH at 1679cm<sup>-1</sup>. The chloro atom in compound(1) was replaced by different aromatic amines to prepare compounds(1A-H) infra red data showed that appearing of –CH<sub>2</sub>NH peak at 2886cm<sup>-1</sup> and disappearing of carbone-chlore peak at 750cm<sup>-1</sup>. Chloro acetyl group in compound(1) was completely converted to thiazolidinone ring system (compound 2) infra red data showed peaks at 1714cm<sup>-1</sup> (C=O),1537m<sup>-1</sup> (C=N-H). Using different aromatic aldehyde, Compound(2) was converted to compounds(2A-H), Infra red data showed that peaks at 1720cm<sup>-1</sup>,1714cm<sup>-1</sup> ,(C=O) ,1550cm<sup>-1</sup>(C=N-H). <sup>1</sup>H-n.m.r. spectra showed peaks at (10.6-10.7 ppm) for N-H proton, CH<sub>3</sub> and CH<sub>2</sub> at (2.2-3.4 ppm), protons of Benzene ring were showed at area (7.2-7.9 ppm) as multiple peaks. The figures from (3-7) state the signals of some prepared compounds and table (2).

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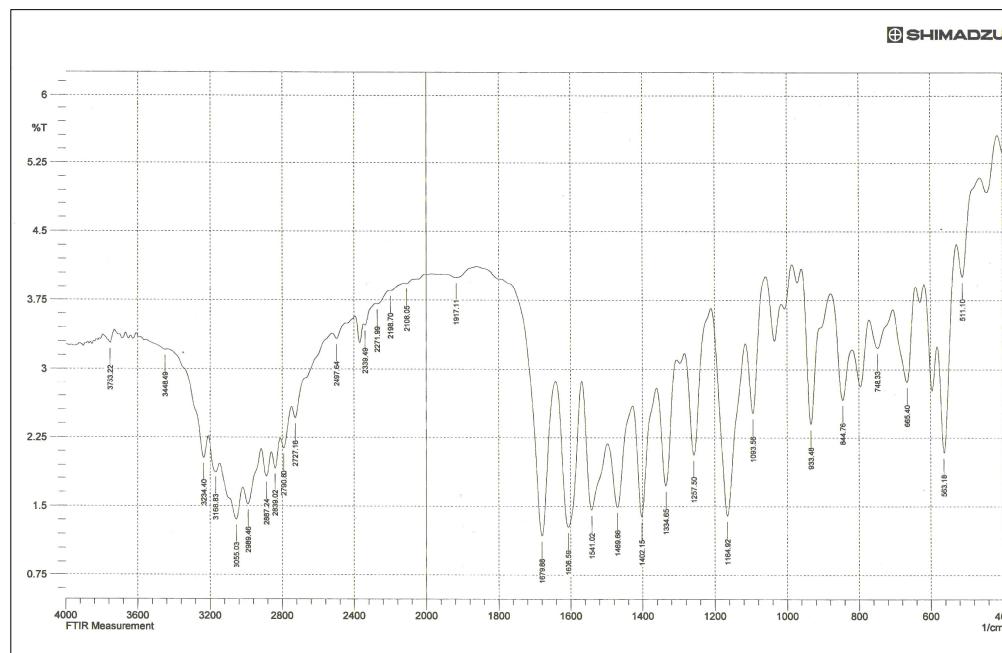


Fig.(1) IR spectrum of synthesized compound (1)

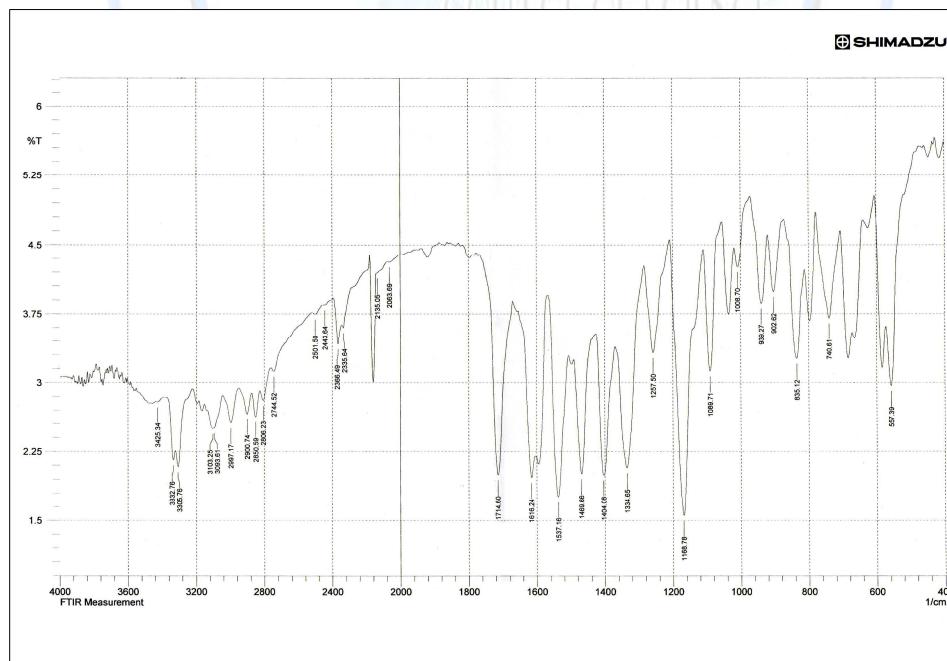


Fig. (2) IR spectrum of synthesized compound (2)

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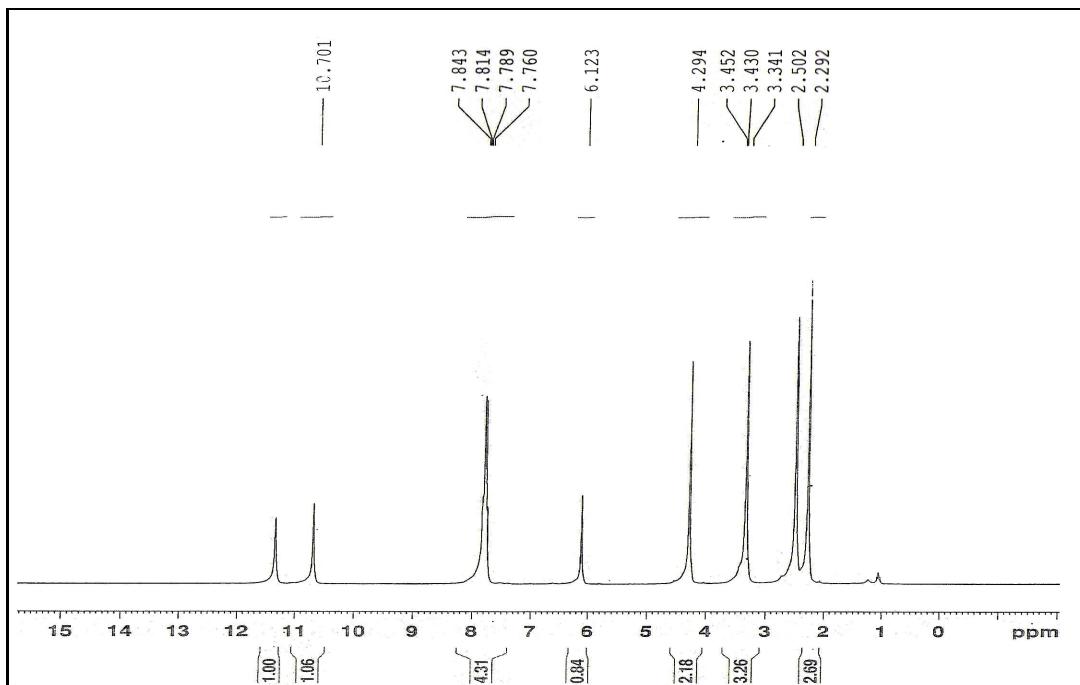


Fig.(3) <sup>1</sup>H-n.m.r. spectrum of synthesized compound (1D)

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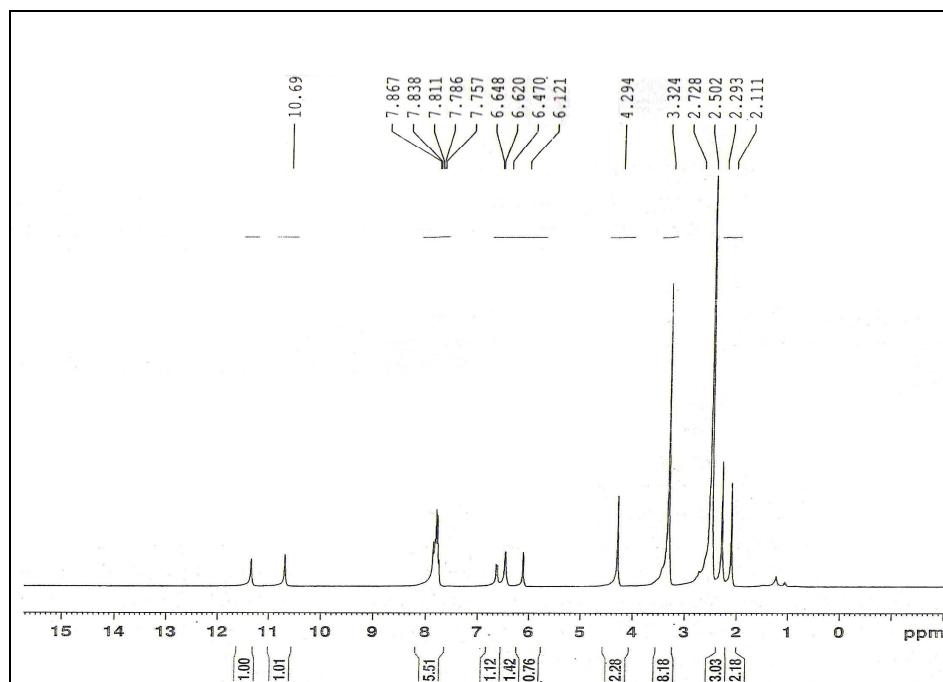


Fig.(4)  $^1\text{H}$ -n.m.r. spectrum of synthesized compound (1H)

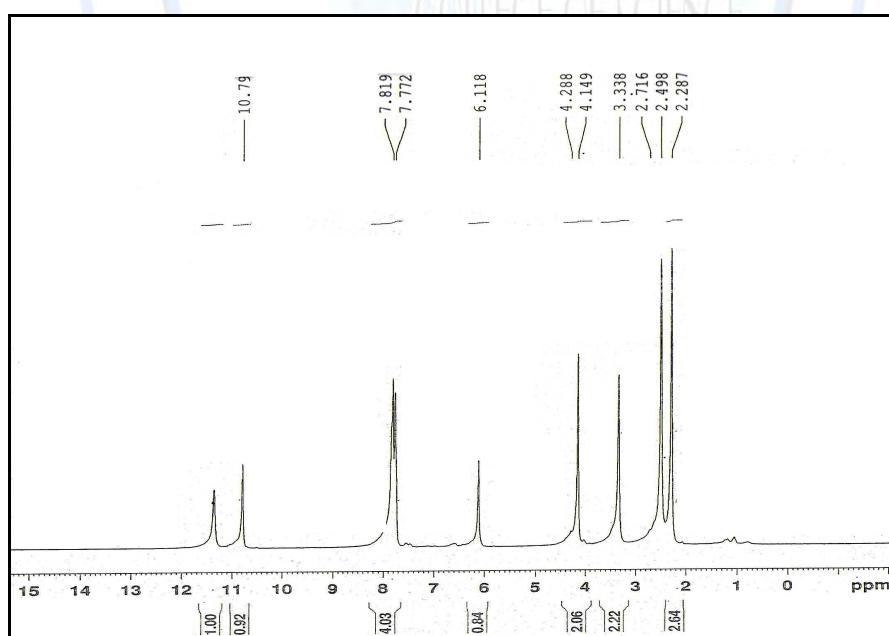
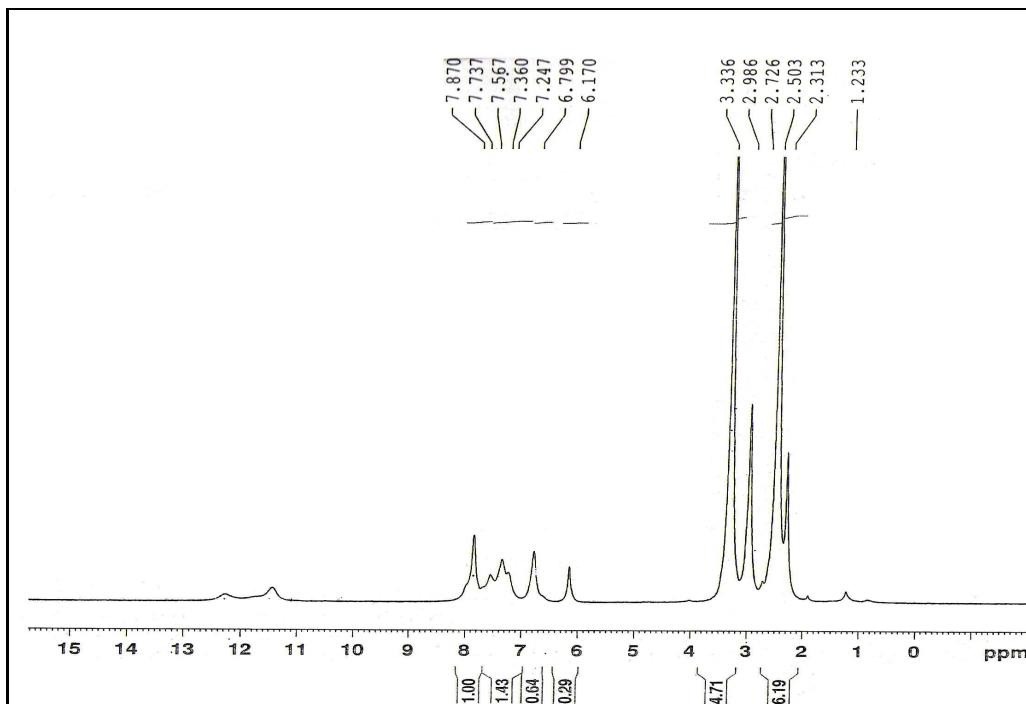


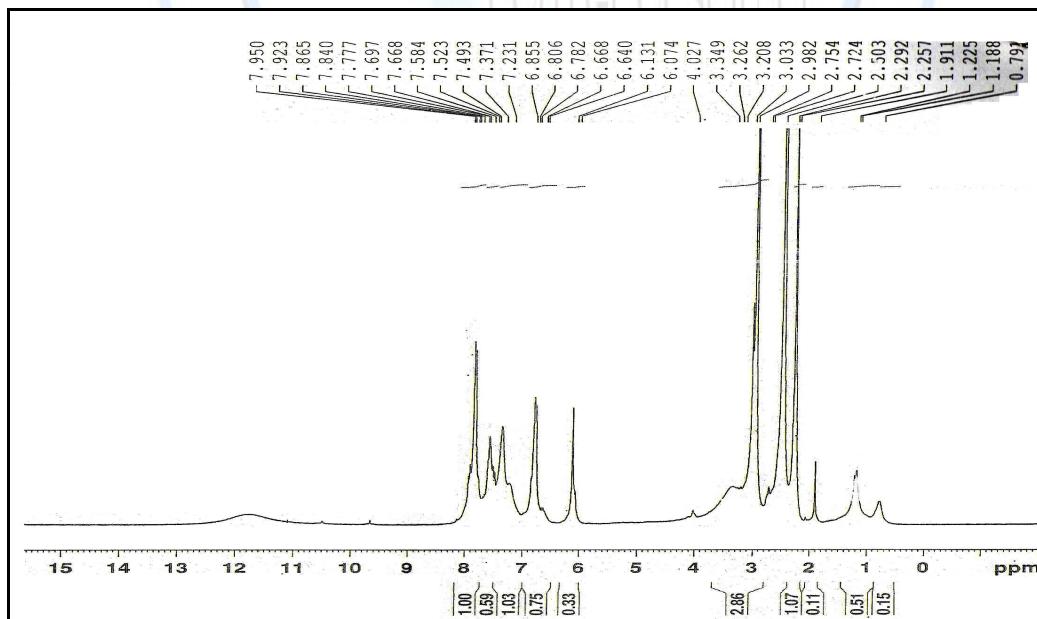
Fig.(5)  $^1\text{H}$ -n.m.r. spectrum of synthesized compound (2)

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**Fig.(6)**  $^1\text{H}$ -n.m.r. spectrum of synthesized compound (2B)

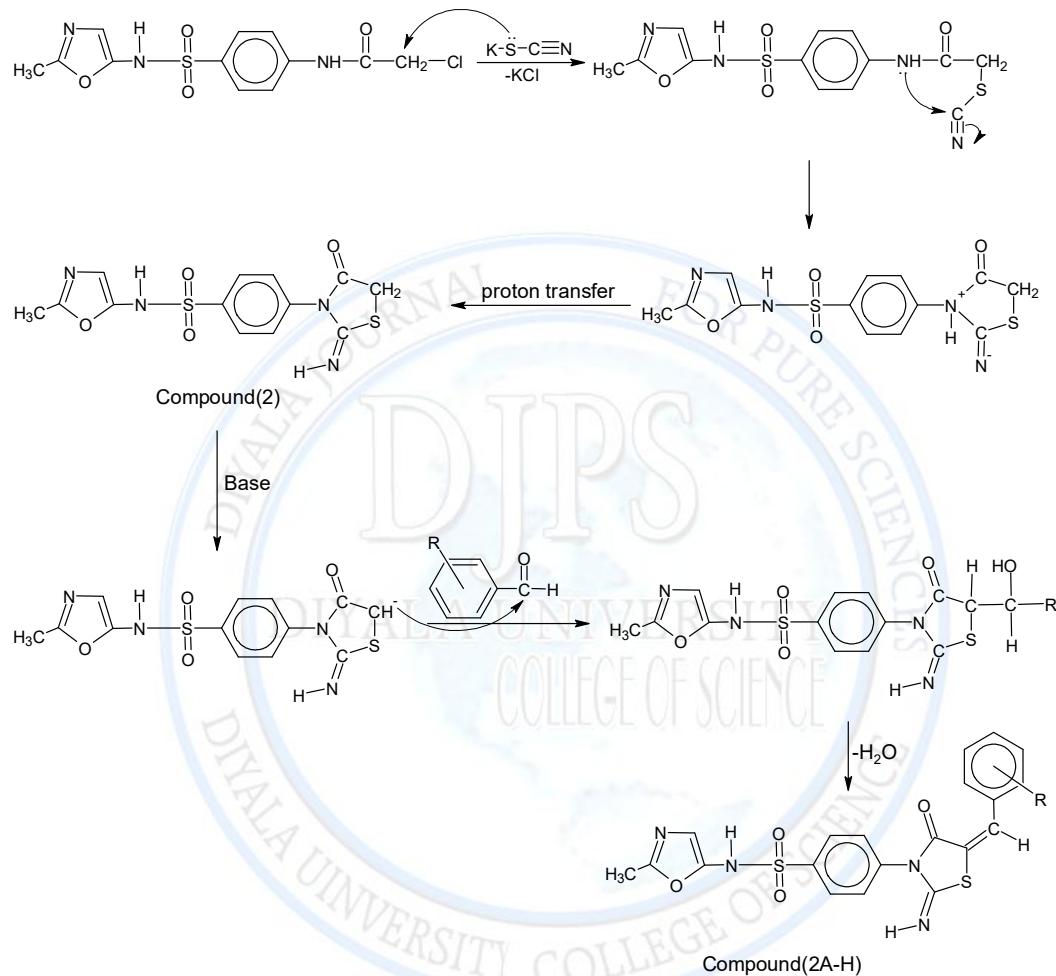


**Fig.(7)  $^1\text{H}$ -n.m.r. spectrum of synthesized compound (2E)**

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**Scheme 2: Mechanism of reaction for compounds (2) and (2A-H).**



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**Table 1. Physical properties of the prepared compounds**

| Compound no. | R                                    | M.P     | Color  | Yield |
|--------------|--------------------------------------|---------|--------|-------|
| 1            | -                                    | 195-197 | white  | 83    |
| 1A           | p-OCH <sub>3</sub>                   | 182-184 | White  | 75    |
| 1B           | p-COOH                               | 175-177 | White  | 77    |
| 1C           | m-NO <sub>2</sub>                    | 114-116 | Yellow | 72    |
| 1D           | o-NO <sub>2</sub>                    | 186-188 | Orange | 70    |
| 1E           | m-OH                                 | 212-214 | Brown  | 76    |
| 1F           | p-OH                                 | 225-227 | Brown  | 74    |
| 1G           | p-Br                                 | 206-208 | White  | 72    |
| 1H           | o-CH <sub>3</sub> ,p-NO <sub>2</sub> | 180-182 | Yellow | 78    |
| 2            | -                                    | 167-169 | Yellow | 70    |
| 2A           | p-OH                                 | 217-219 | Brown  | 67    |
| 2B           | p-N(CH <sub>3</sub> ) <sub>2</sub>   | 247-249 | Red    | 55    |
| 2C           | p-OH,m-OCH <sub>3</sub>              | 204-206 | Yellow | 62    |
| 2D           | p-Cl                                 | 253-255 | White  | 60    |
| 2E           | p-OCH <sub>3</sub>                   | 264-266 | White  | 56    |
| 2F           | m-OH                                 | 241-243 | Red    | 64    |
| 2G           | o-Cl                                 | 244-246 | White  | 58    |
| 2H           | O-CH <sub>3</sub>                    | 248-250 | White  | 53    |

**Table 2. <sup>1</sup>H.N.M.R signals in ppm for some prepared compounds**

| Compound no. | (-CH <sub>3</sub> ) ppm | (-CH <sub>2</sub> ) ppm | (-CH) <sub>Ar</sub> ppm | (N-H) ppm |
|--------------|-------------------------|-------------------------|-------------------------|-----------|
| 1D           | 2.29-2.50               | 3.34-4.29               | 7.76-7.84               | 10.70     |
| 1H           | 2.11-2.72               | 3.32                    | 7.75-7.86               | 10.69     |
| 2            | 2.28-2.71               | 3.33                    | 7.77-7.81               | 10.79     |
| 2B           | 2.31-2.98               | 3.33                    | 7.24-7.87               | 10.78     |
| 2E           | 2.52-2.75               | 3.03-3.34               | 7.23-7.95               | 10.68     |

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**Table 3. Wave numbers in  $\text{cm}^{-1}$  of I.R spectrum for prepared compounds**

| Compound no. | $(\text{C}-\text{H})_{\text{aliph}}$ | $(\text{C}-\text{H})_{\text{Ar}}$ | $(\text{C}=\text{C})_{\text{Ar}}$ | $(\text{C}=\text{N})$ | $(\text{C}=\text{O})$ | $(\text{N}-\text{H})$ |
|--------------|--------------------------------------|-----------------------------------|-----------------------------------|-----------------------|-----------------------|-----------------------|
| 1            | 2886                                 | 3066                              | 1550                              | 1608                  | 1670                  | 3234                  |
| 1A-H         | 2882-2889                            | 3075-3098                         | 1540-1560                         | 1612-1620             | 1665-1668             | 3165-3223             |
| 2            | 2908                                 | 3093                              | 1537                              | 1616                  | 1716-1720             | 3206                  |
| 2A-H         | 2885-2997                            | 3095-3103                         | 1530-1542                         | 1610-1625             | 1714-1716             | 3220-3340             |

**References**

1. M. Vigorita, R. Ottana, F. Monforte. *Bio.med. chem. let.* 11,2001,279.
2. M. Diurno, M. Mazzoni, O.Izzo. *Il Farmaco.* 52,1997,237.
3. N. Ergene, G. capan. *Il Farmaco.* 49,1998,449.
4. R. Sharma, D. Kumar. *J. Indian. Chem. Soc.* 77,2000,492.
5. T. Kato, T. Ozaki, N. Ohi. *Tetrahedron.* 109,1999,3963.
6. A. Ameya, R. Alandine. *Arkivoc.xxi.* 2007,148-155.
7. P. Butvinp, M. Mitkova, E. Harrnek. *Chem.pap.* 56,2002,174-177.
8. N. Ali, S. Jamal, N. Davood. *Turk. j. chem.* 30,2006,619-629.
9. D. Bryna, S. Naim, S. Paul. *J. American. Chem. Soc.* 8,2008,10.
10. Z. Turget, C. Yolacan, F. Aydogan, E. Bagdatali. *Molecules.* 12,2007,2151- 2159.
11. S. Bouzoura, Y. Beutarzi, R. Kaoua, B. Nedjar. *J.Org. Commun* 3,2010,8-14.
12. K. Mistry, K. Desai. *E. J. of Chem.* 1,2004,189-193.
13. R. Patal, K. desai, K. Chikhalia. *Indian J. of Chem.* 54b,2006,773-778.
14. A. Gursoy, T. Iyikosker, N. Terzioglu, G. Otuk. *Turk. J. Pharm. Sci.* 2,2005,1-10.
15. A. Niazi, S. Jameh, D. Nori. *Turk. J. Chem.* 30,2006,619-628.
16. E. Pinho, *Cur. Org. Chem.* 9,2005,925.
17. P. Wipf, *Chem. Rev.* 95,1995,2115-2134.
18. B. lipshutz. *Chem. Rev.* 86,1986,795-829.
19. P. Wipf, C. Miller. *J. Org. Chem.* 58,1993,3604-3606.



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20. C. Mody, K. Doyle, *Pg. Hetero. Chem.* 9, 1997, 1-16.
21. L. Susy, M. Claudio, F. Rui, P. Teresa, *J. of Molecular structure* .919, 2009, 47-53.
22. Z. Justyna, O. Dorota, Z. Zofia, *Act. Polo. Pharm. drug research.* 65, 2008, 229-233.
23. M. Jiang, L. Chen, H. Xian, *J. Chin. Chem. Let.* 15, 2004, 143-144.
24. K. Manasi, T. Jetze, *J. Org. Chem.* 70, 2005, 4211-4213.
25. C. Usifon, L. Okunrobo, *Afr. J. of Sci. and Tech.* 2, 2001, 47-50.
26. Z. Zhan, L. Feng, Z. Shi, *J. Chine. Chem. let.* 12, 2001, 947-950.
27. K. Mohamad, P. Venug, R. Mohana, *Let. in drug. design and discovery* 6, 2009, 21-28.
28. A. Nuha, *Thesis, Al-nagah University.* 2010.
29. A. Ameya, R. Nandini, *Molecules*, 12, 2007, 2467-2477.
30. R. Lakhan, O. Singh, *J. Ind. Chem. Soc.*, 61, 784, 1984.
31. I. Paltork, M. Khosropour, R. Hojati, *J. Chemical Monthly* , 183, 663, 2007.
32. T. Katsura, Y. Inoue, Y. nishior, *J. Chem. Pharm. Bull.* (Tokyo), 40, 1818, 1992.