

Synthesis and Liquid Crystalline Study of 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzylidene-4''-n'-alkoxy aniline

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Abstract

The synthesis and study mesomorphic properties of a new series of 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzylidene-4''-n'-alkoxy aniline (DC₇A₁₋₁₀) Have been reported. The structures of the prepared compounds were confirmed by FT-IR, ¹H-NMR and elemental analysis (CHN). The mesomorphic properties were studied by differential scanning calorimetry (DSC) and polarizing optical microscopy (POM) measurements. All compounds of the series exhibited nematic (N) phases but the smectic phases are appear in (DC₇A_{3-A10}). The first two series (DC₇A₁, DC₇A₂) display only a nematic phases whereas the highest homologues (DC₇A_{3-A10}) exhibit enantiotropic dimorphism Nematic (N) and smectic (S) phases (the DC₇A₆ is only compound exhibit S_c and S_A phases). The liquid crystalline behavior has been analyzed in terms of structural property relationship. The mesomorphic properties and thermal stabilities of the present series (DC₇A₁₋₁₀) were compared with those structurally related homologous series.

Keywords: Mesomorphism, mesogen, enantiotropy, nematic, smectic.

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تحضير و دراسة الصفات البلورية السائل لمشتقات (2,4-بس (4'-ن-هبتايلوكسي بنزوايلوكسي)

بنزايلايدين-4''-ن'-الكوكسي انيلين

ولى محمود حمد

قسم الكيمياء - فاكلتى العلوم والصحة - جامعة كويه

الخلاصة

تم تحضير ودراسة الصفات البلورية السائلة لسلسلة جديدة 2,4-بس (4'-ن-هبتايلوكسي بنزوايلوكسي) بنزايلايدين-4''-ن'-الكوكسي انيلين (DC_{7A1-10}). شخّصت المركبات المحضرة جميعا بواسطة اشعه تحت الحمراء (FT-IR)، الرنين النووي المغناطيسي (¹H-NMR) وكذلك تحليل العناصر (CHN). ودرست الصفات البلورية السائلة بواسطة المسح المسعري التفاضلي (DSC) وكذلك المجهر ذات صور المستقطب (POM). جميع المركبات المحضرة الجديدة أظهرت الطور البلوري السائل النيماتى (N)، لكن الطور البلوري السائل السمكتى (S) فقد أظهرت مع المركبات (DC_{7A3-A10}) ان المركبين ذات تراكيب (DC_{7A1-DC7A2}) أظهرت فقط الطور البلوري السائل النيماتى (N)، ولكن المركبات ذات عدد الكاربون عالي (DC_{7A3-DC7A10}) أظهرت الصفات البلورية الاناثيوتروبي (السمكتى وكذلك النماتى). ولكن المركب الوحيد ضمن السلسلة المحضرة ذات تركيب (DC_{7A6}) أظهرت مع الطور النيماتى الطور السمكتى (S_A & S_C). ولوحظ هناك علاقة قوية بين تراكيب الكيمياوية للمركبات وكذلك الاطوار البلورية السائلة. ودرس الاستقرار الحراري وجد بان ظهور الاطوار البلورية السائلة وقورنت مع مركبات محضرة ذات تراكيب مشابهة لهذة السلسلة.

الكلمات المفتاحية: البلورات السائلة، ميزوجين، الطور السمكتى، الطور النيماتى.

Introduction

Aromatic azomethine ester comprising of different polarity of substituent has been known to either promote or suppress the mesomorphic properties [1, 2]. The molecular breadth is usually derived from a lateral substitution. Lateral substitution plays an effective role in the mesomorphic properties at the compound [3]. Aziz H. J has been synthesized some new Schiff base diester compounds with the general structure [4] (Figure 1).

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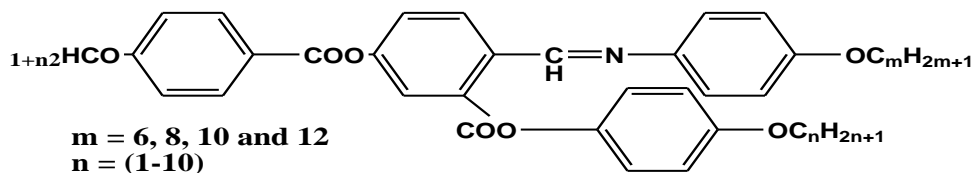


Figure 1

The mesomorphic behavior of this Schiff base compound show the (N), (Sc) and (SI) mesophases. In this work, we describe the synthesis and mesomorphic properties of homologous series (DC₇A₁₋₁₀) (Figure 1I).

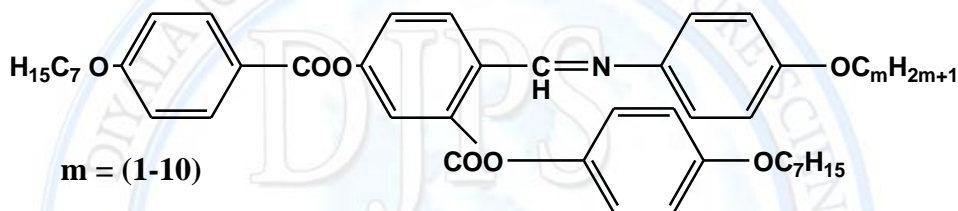


Figure 1I

Experimental

1- Materials

4-hydroxyacetanilide, 2,4-dihydroxybenzaldehyde, tetrahydrofuran(THF), triethylamine (Et₃N), benzene, absolute ethanol, thionyl chloride, potassium hydroxide, 4-hydroxy benzoic acid, hydrochloric acid, sodium carbonate and glacial acetic acid were purchased from Sigma-Aldrich. The chemicals were used as received.

2- Instrumentations

Elemental analysis (HN) was carried out using a Perkin-Elmer model 2400 instrument. ¹H-NMR spectra obtained by (Bruker 500MHz) using (TMS) as the internal standard. Infrared spectra were recorded by FT-IR Shimadzu 8000 instrument, in wavenumber (4000 – 400 cm⁻¹), using KBr pellet. The optical behavior observations were made using Olympus BX40 microscope equipped with a Leitz Laborlux 12 puls hot stage and PR 600 controller. Photographs of texture were obtained using a camera model PM-10 AD made Olympus. The transition temperatures

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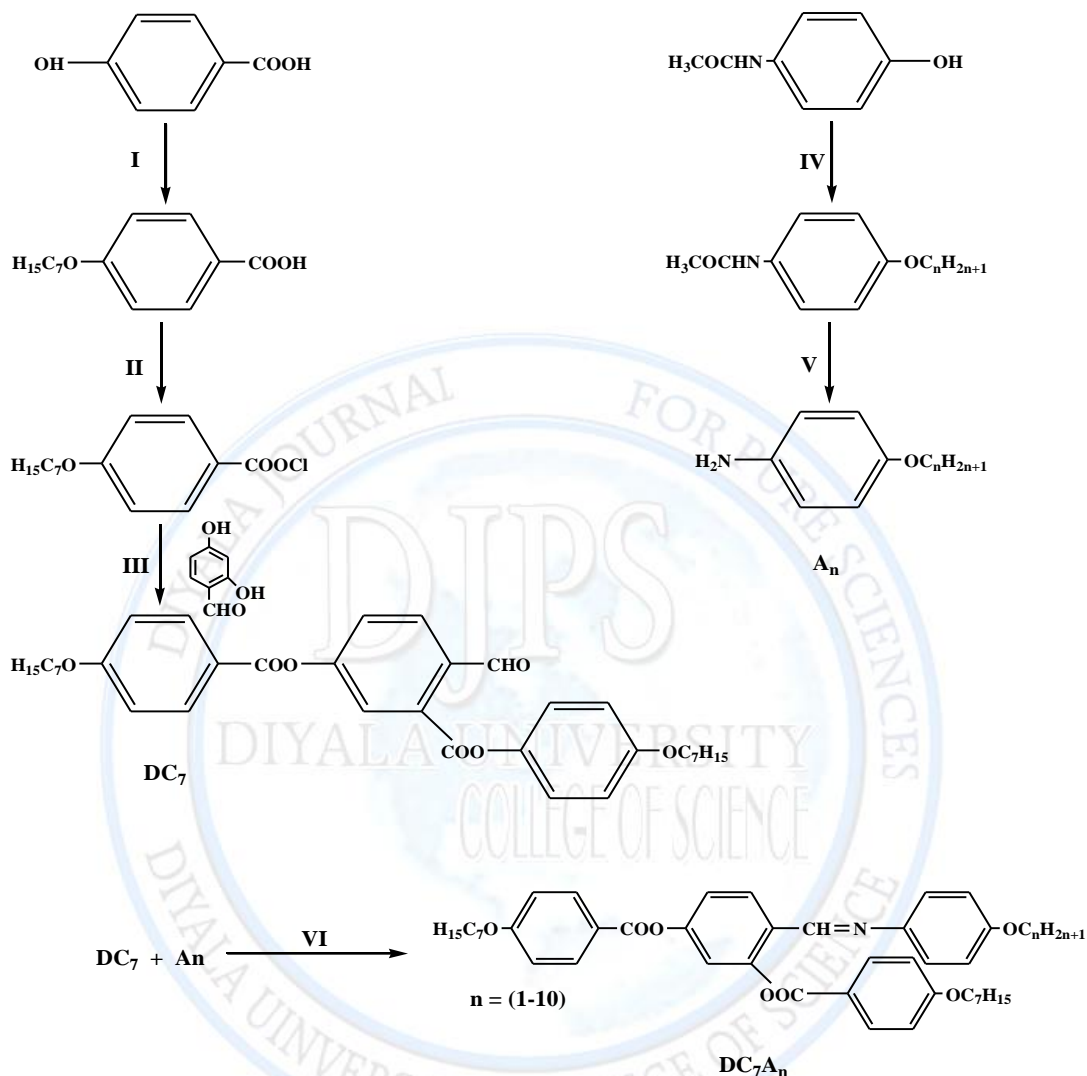
and enthalpies were investigated using differential scanning calorimetry DSC (TA instruments Q 1000 DSC), ramp rate: 10 °C/min under N₂ atmosphere.

3- Synthesis

4-n- heptyloxy benzoic acid, 4-n- heptyloxy benzoyl chloride, 4-n- alkoxy acetanilides and 4-n- alkoxy anilines derivatives were prepared according to the refs. [5, 6]. 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzaldehyde was synthesized by the method of Pugh and Schrock and Zhou et. al. [7, 8]. Synthesis of series 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzylidene-4''-n-alkoxy aniline (DC₇A₁₋₁₀). The mixture of 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzaldehyde (0.5 mmol) and 4-n-alkoxy pceaniline (0.5 mmol) and a few drops of glacial acetic acid as catalyst in (10ml) of an absolute ethanal was refluxed for (6-7) hours. The yellow precipitate was formed during the cooling of mixture. Then it was filtered off, washed (2-3) times with cold ethanol. Finally, the obtained solid (DC₇A_n) was recrystallized from an absolute ethanol until the transition temperature remains constant. Yield (69-72%). The synthetic route used for the preparation of 2,4-bis(4'-n-heptyloxybenzoyloxy)-benzylidene-4''-n-alkoxy aniline (DC₇A_n) is shown in (scheme I).

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- (I) $\text{C}_7\text{H}_{15}\text{Br}$, KOH, Ethanol/ H_2O , reflux, 25h., SOCl_2 , C_6H_6 , reflux, 1h. (II) THF, Et_3N , 5°C , 24h.
 (III) C_nH_{2n} , Br, KOH, Ethanol, reflux, 24h. (IV) KOH, Ethanol, reflux, 4h. (V) Ethanol, glacial acetic acid, reflux, 7h.

Scheme I: synthetic route for compounds DC_7A_n

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Results and Discussion

1- Characterization

FT-IR (KBr, cm^{-1}) 2960-2850 cm^{-1} ($\nu_{\text{C-H}}$ aliphatic); 1740-1725 cm^{-1} ($\nu_{\text{C=O}}$ ester); 1605-1625 cm^{-1} ($\nu_{\text{C=N}}$ azomethine); 1105-1250 cm^{-1} ($\nu_{\text{C-O-C}}$ aryl ether); 619-880 cm^{-1} ($\nu_{\text{C-H}}$ aromatic).

$^1\text{H-NMR}$ (DMSO- d_6), ppm; $\delta=8.61$ (s, 1H, $-\text{CH=N-}$); $\delta=6.87$ -8.13(dd, 15H, rings); $\delta=3.99$ -4.09(t, 6H, $-\text{CH}_2\text{O -}$); $\delta=1.40$ -3.3(m, 15H, 6 $-\text{CH}_2-$ and $-\text{O-C-CH}_3$) and $\delta=0.89$ -0.92(t, 6H, $-\text{CH}_3$).

The elemental analysis:

Table 1: Elemental analysis of all compounds

No.	Compounds	Molecular weight	% Found (Calculated)		
			C	H	N
1	DC ₇ A ₁	679.2	74.32(74.22)	7.01(7.21)	1.96(2.06)
2	DC ₇ A ₂	693.3	74.38(74.74)	7.16(7.35)	1.82(2.02)
3	DC ₇ A ₃	707.4	74.81(75.24)	7.30(7.49)	1.80(1.98)
4	DC ₇ A ₄	721.0	75.32(75.72)	7.54(7.62)	1.79(1.94)
5	DC ₇ A ₅	735.0	76.01(76.19)	7.78(7.75)	1.80(1.90)
6	DC ₇ A ₆	749.0	76.23(76.63)	7.82(7.87)	1.74(1.86)
7	DC ₇ A ₇	763.0	76.59(76.53)	8.01(7.99)	1.84(1.83)
8	DC ₇ A ₈	777.0	76.80(76.96)	8.12(8.10)	1.72(1.80)
9	DC ₇ A ₉	791.0	77.12(77.37)	8.18(8.21)	1.68(1.76)
10	DC ₇ A ₁₀	805.0	77.82(78.01)	8.29(8.32)	1.62(1.73)

2- Mesomorphic properties:

The mesophases of the present series (DC₇A₁₋₁₀) were observed by polarized optical microscope (POM) during heating and cooling cycles. The thermal behaviors obtained by DSC are consistent with the data observed by (POM). The data obtained by DSC and POM analysis of the present series (DC₇A₁₋₁₀) are summarized in Table (2).

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Table 2: Transition temperatures and total enthalpy changes of (DC₇A_n)

No.	Compounds	Transition temperatures °C (ΔH) (kJ/mol)
1	DC ₇ A ₁	cr 257 N > 300 I (25.76)
2	DC ₇ A ₂	cr 121 N 246 I (18.23)
3	DC ₇ A ₃	cr 57 SmC 72 N 101 (13.89) I
4	DC ₇ A ₄	cr 52 SmC 62 N 126 (15.76) I
5	DC ₇ A ₅	cr 121 SmC 138 N 164 (12.20) I
6	DC ₇ A ₆	cr 103 SmC 131 SmA 149 (15.40) N 182 I
7	DC ₇ A ₇	cr 82 SmC 127 N 150 (14.80) I
8	DC ₇ A ₈	cr 60 SmC 84 N 115 (22.89) I
9	DC ₇ A ₉	cr 107 SmC 124 N 140 (19.52) I
10	DC ₇ A ₁₀	cr 116 SmC 121 N 126 (9.53) I

cr=Crystal; N=Nematic; S_mA=Smectic A; S_mC=Smectic C; I=Isothipic transition.

All the members of series DC₇A_n exhibit enantiotropic mesophases. The DC₇A₁ and DC₇A₂ are homologues are purely enantiotropic mematogenic. With increasing length of Alkoxy chain of DC₇A₃ to DC₇A₁₀ the enantiotropic smectic (C) phase was observed and accompanied the nematic phase. The nematic phase start from n=1 and persists up to the last number of the series. The persisting of nematic phase indicates that the net effect imparted due to the end-to-end cohesive forces in this series is similar as all the other molecular forces are the same. The increasing of carbon chain (n>2) leads to the formation of smectic mesophase due to enhances lateral attraction as compared with terminal attractions, which facilitates lamellar packing that required for the formation of the smectic mesophase [9]. The appearance of S_mA phase for compound (DC₇A₆) was understandable and that could be that the three alkoxy terminal groups with 7, 7 and 6 carbon atoms give rise to more planner and polarizable geometrical structure of DC₅A_n [10]. The plot of transition temperatures against the number of carbon atoms in the alkoxy chain (Figure 2). Showed the relation between carbon number (n) of alkoxy group and transition temp. of mesophases.

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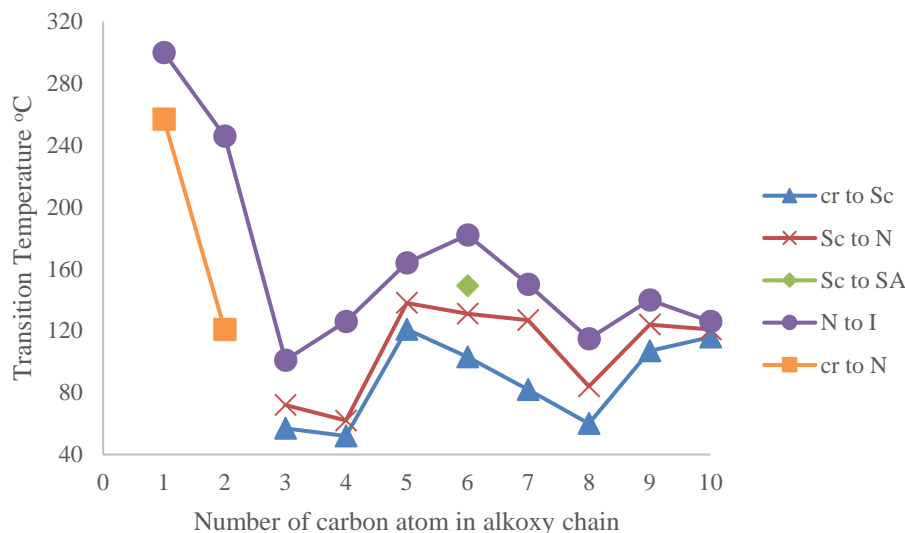


Figure 2: Dependence of transition temperature on the increasing number of carbon atom (n) in the terminal alkoxy chains for the DC₇A₁₋₁₀ series of compound

The mesophases exhibited by compounds of series DC₇A₁₋₁₀ were identify according to their optical textures, which were observed by POM, using the classification systems reported by Demus and Richter [11] and Dierking [12]. The optical photographs of DC₇A₃ are depicted in (Figure 3) as a representative illustration,

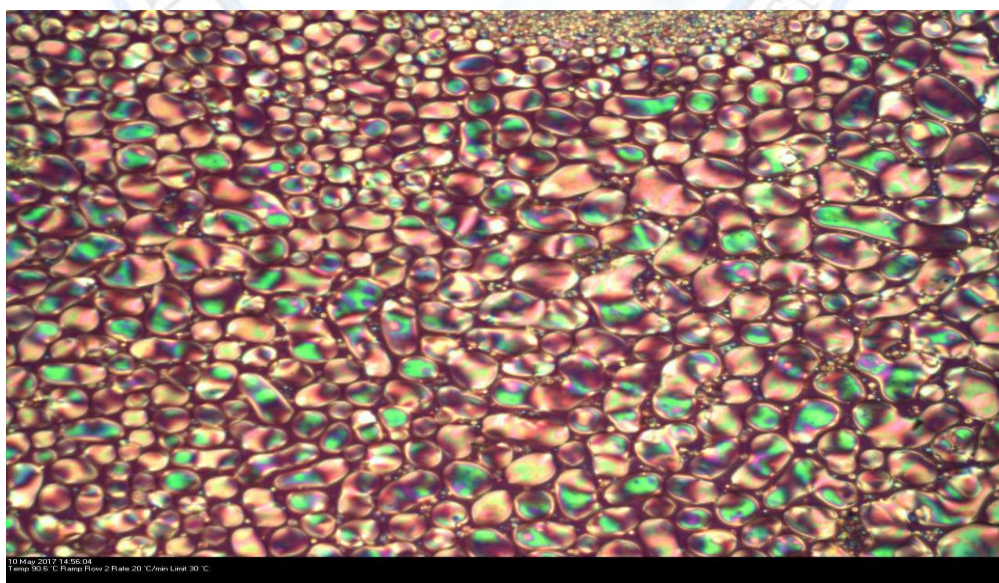


Figure 3: The Nematic droplet for the compound DC₇A₃

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nematic droplet obtained in cooling. Upon cooling the isotropic liquid of each compound, nematic droplets appeared and Schlieren texture with characteristic two- and four- Brush singularities. When the compound further cooling (Figure 4) the thread like nematic was appeared.

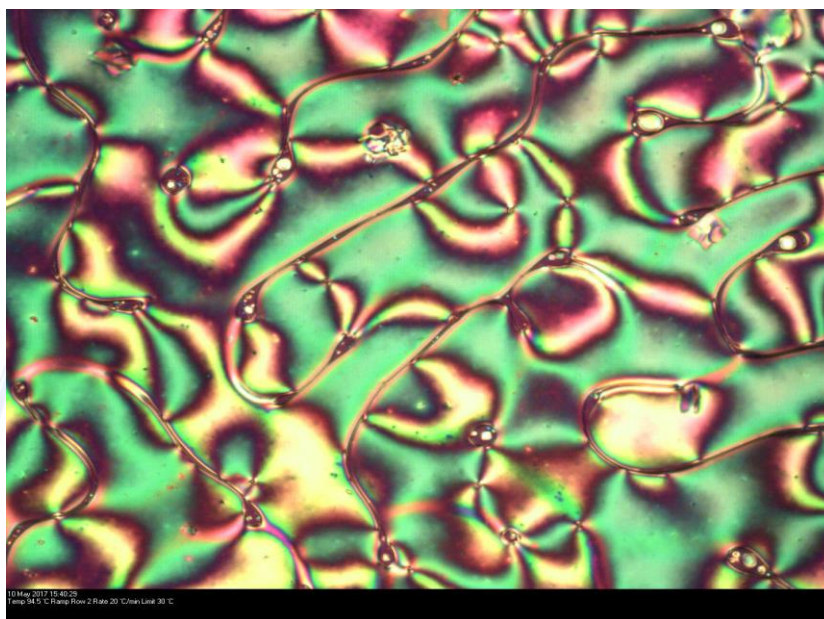


Figure 4: The Thread-like nematic of compound DC₇A₃

The typical S_mA for DC₇A₆ was appeared by cooling (Figure 5) as a broken focal conic S_mA.

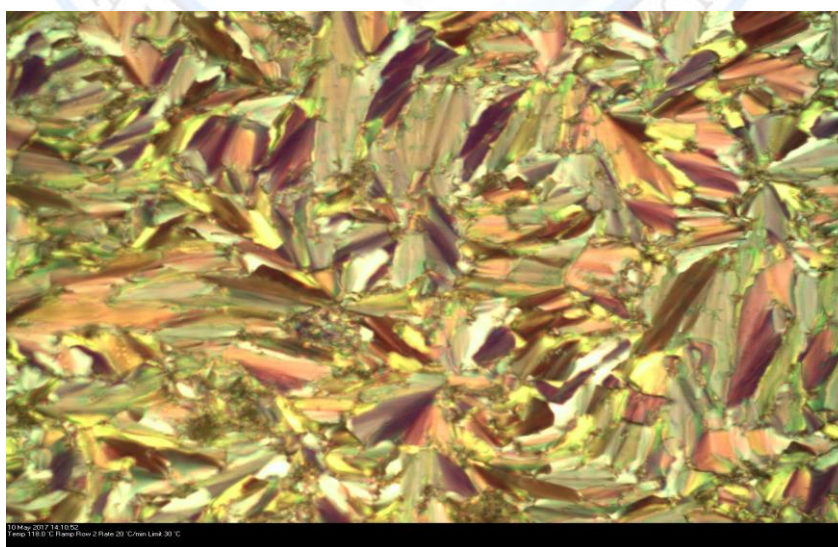


Figure 5: The Broken focal conic smectic A of compound DC₇A₆

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When the compound DC₇A₈ cooling from isotropic phase the thread like S_mC appeared (Figure 6).



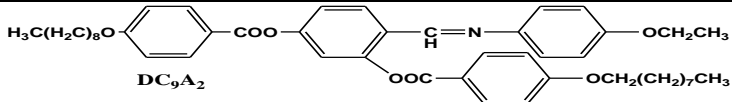
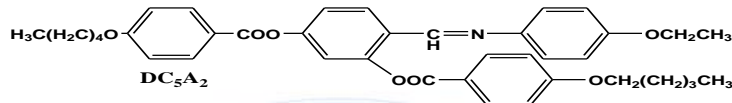
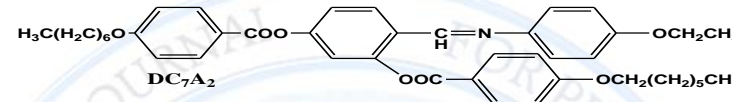
Figure 6: The Thread-like smectic C of compound DC₇A₈

It was found that there is close relationship between mesomorphism and molecular constitution of mesomorphic compounds [13, 1, 14]. This relationship could be correlated with the molecular constitution of these compounds by thermal stability, which is measure of mesomorphism. Table (3) shows the comparison of mesomorphic properties and thermal stability of the three series (present series, DC₇A_n and two presence series DC₉A_n [15] and DC₅A_n [10]) show a very similar mesomorphic behavior displaying nematic mesomorphic. The members in all series with short alkoxy chains (n=2) show a nematic mesophase.

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Table 3: Comparison between transition temperatures, mesomorphism and average thermal stabilities for DC₉A₂, DC₅A₂ and DC₇A₂

1.	 DC ₉ A ₂		
2.	 DC ₅ A ₂		
3.	 DC ₇ A ₂		
No.	Compounds	Transition temperatures °C	ΔT
1	DC ₉ A ₂	cr 127 N 167 I	40 °C
2	DC ₅ A ₂	cr 109 N 216 I	107 °C
3	DC ₇ A ₂	cr 121 N 246 I	125 °C

Conclusions

A new series of 2,4-bis (4'-n-heptyloxybenzoyloxy)-benzylidene-4''-n'-alkoxy aniline (DC₇A_n) have been prepared by changing the terminal alkoxy chain length (n=1-10). The chemical structures of the series were identified by IR, ¹H-NMR and elemental analysis. The results are in agreement with the considered molecular structure. The liquid crystal properties and optical textures of the compounds were studied by POM and DSC. The change in terminal alkoxy chain length has pronounced effect on the mesomorphic behavior. The first two homologues (DC₇A₁, DC₇A₂) exhibited only N phases but the others (DC₇A₃, DC₇A₁₀) display dimesomorphism N and S_mC phases. But with compound (DC₇A₆) the other S_mA phase was observed. The mesomorphic properties and thermal stabilities of the present series is compared with those of structurally related homologous series.

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